# Material Safety Data Sheet for Lithium Ion Cells

# Section 1 Chemical Product and Company Identification

Trade name: Lithium ion cells

Model: All 18650 models with capacity less than or equal to

3Ahr.

Nominal voltage: 3.6V or 3.7V

# Section 2 CompositionIInformation on Ingredients<1,

Chemical name	Molecular formula	CAS No.	Classification (approximate )
Lithium transition metal oxide">	LixM0 <sub>2</sub>	proprietary	depend on model
Carbon (graphite)	c	7440-44-0	20%
Aluminum	Al	7429-90-5	3.5%
Copper	Cu	7440-50-8	8%
Steel	Fe	7439-89-6	I
Electrolyte	/	proprietary	Flammable liquid

#### Notes:

- 1) Not every product contains all of these materials
- 2) M means a combination of Co, Ni or Mn. This component may consist of a mixture of compounds, each of which may contain these elements.

### **Section 3 Hazards Identification**

Lithium ion cells are not hazardous when used according to the instructions of the manufacturer under normal conditions. In case of abuse, there is a risk of rupture, fire, heat, or leakage of internal components, which could release hazardous materials.

#### Section 4 First-aid Measures

Lithium ion cells are not hazardous under normal circumstances. In case of fire or rupture, the leakage of internal hazardous substance and formation of hazardous substance may occur, and the following measures should be taken in case of contact with these:

Eyes: Check for and remove any contact lenses. Immediately flush with plenty of clean water for at lest 15 minutes, seek medical assistance.

Skin: Immediately flush with plenty of clean water for 15 minutes; seek medical assistance if reaction is severe.

Inhalation: Remove to fresh air immediately, seek medical assistance, and ventilate the contaminated area.

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Ingestion: Rinse mouth with clean water immediately. Make the victim vomit and seek medical assistance.

# **Section 5 Fire-fighting Measures**

Extinguish with plenty of water, dry powder extinguishers, sands or earth. Combustion products and decomposition products include: carbon monoxide, carbon dioxide, hydrogen fluoride, phosphorus fluoride.

#### Section 6 Accidental Release Measures

When leakage of cells happens, liquid could be absorbed with sands, earth or other inert substance, and the contaminated area should be ventilated.

## Section 7 Handling and Storage

Handling Precautions:

Do not short positive and negative terminals by contact with conductors. Do not overheat or incinerate. Do not open, puncture, crush or deform cells

Storage: Store and use away from heat, sparks, open flame, or any other ignition source. Store in a cool, dry environment (less than 35 °C, less than 85% RH).

## Section 8 Exposure Controls/Personal Protection

There is no protection required under normal conditions. In case of leakage ventilation is required. Respirator, eye protection, protective gloves and protective clothes are required when dealing with fire and leakage.

#### Section 9 Physical and Chemical Properties

Cells are not single chemical material; there are no specific physical and chemical properties such as melting point and boiling point.

#### Section 10 Stability and Reactivity

Cells are stable under normal conditions. The following substance may appear in case of fire or leakage: organic carbonate, hydrogen fluoride, carbon monoxide, carbon dioxide, phosphorus fluoride.

#### **Section 11 Toxicological Information**

Cells are not hazardous when used properly. In case of fire or leakage combustion and decomposition products may cause irritation and toxicity to skin, eye and respiratory systems. Toxicity data of some substance is listed:

Hydrogen fluoride:

Extremely toxic, May be fatal if inhaled or ingested. Readily absorbed through the skin contact may be fatal. Possible mutagen. LCLO: 50 ppm/30m (human beings), LC50: 1276 ppm/1h crats).

Carbon and graphite:

Slightly hazards in case of skin contact (irritant), ingestion, inhalation, which will cause chronic

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damage to upper respiratory tract and cardiovascular system.

Copper

Dust may cause respiratory irritation.

LD50: 3.5 mg kg· (mouse).

# **Section 12 Ecological Information**

There is no influence on ecology or environment when used and disposed of properly.

## **Section 13 Disposal**

Discarded cells should not be treated as ordinary trash. Recycling is recommended and is required by law in many jurisdictions. Do not incinerate. Leaking or damaged cells should be treated as chemical waste. Packaging is normally not contaminated by cells.

# **Section 14 Transport Information**

The following regulations apply to the transport of Lithium Ion cells worldwide:

- 1) UN Recommendations on the Transport of Dangerous Goods according to which Lithium ion cells are assigned UN ID#3480, Class 9, Packing group II
- 2) International Air Transport Association (IATA) Dangerous Goods Regulations (DGR)
- 3) International Maritime Organization (IMO) International Maritime Dangerous Goods (IMDG) Code
- 4) IATA DGR 54th Edition for transportation
- 5) Transport fashion: by air, by sea

Lithium ion batteries containing no more than 1.5g/cell and 8g/battery pack equivalent lithium can be shipped as 'non-dangerous goods' providing the cells have passed the required tests according to UN38.3 (UN Manual of Tests and Criteria, Part III, Subsection 38.3). In this case packing instruction 965-967 of IATA-DGR and special provisions 188 of IMO-IMDG code also apply.

## **Section 15 Regulatory Information**

For shipping regulations see section 14.

## **Section 16 Other Information**

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Serial Number	Change item	Change Content	PIC	Date
1	NO	first edition		2011-3-15
2	Add new information in Section 14	<ul><li>4) IATA DGR 53nd Edition for transportation</li><li>5) Transport fashion: by air, by sea</li></ul>		2012-01-04
3	Section 14	1ATA DGR 54th Edition for transportation		2012-12-27
4	Section 1 Chemical Product Section2Composition/Information on Ingredients	Nominal voltage: 3.6V or 3.7V add classification (approximate )		2013-1-13

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